Rational Design of 2*H*-chromene-based Antiphytovirals that Inhibit Virion Assembly by Out-Competing Virus Capsid-RNA Interactions

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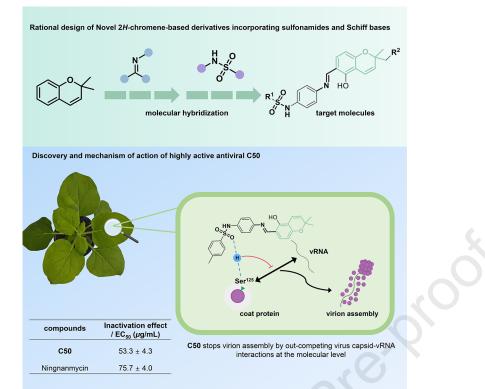
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2	Assembly by Out-Competing Virus Capsid-RNA Interactions
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20	SUMMARY
20	SUMMARI

Although the determination of the structural basis of potato virus Y (PVY) coat protein (CP) 21 22 provides the possibility for CP-based antiviral drug design, the role of many specific residues on CP 23 in regulating virion pathogenicity is largely unknown, and fewer small-molecular drugs have been discovered to act on these potential sites. In this study, a series of derivatives of 2,2-dimethyl-2H-24 25 chromene are rationally designed by employing a molecular hybridization strategy. We screen a case of phytovirucide C50 that could form a stable H-bond with Ser¹²⁵ of PVY CP to exert antiviral 26 properties. Ser¹²⁵ is further identified to be crucial for CP-vRNA interaction, enabling PVY virion 27 28 assembly. This interaction can be significantly inhibited through competitive binding with 29 compound C50. The study enhances our understanding of anti-PVY drug mechanisms and provides 30 a basis for developing new CP-targeting virus particle assembly inhibitors. 31

32 INTRODUCTION

33 Potato (Solanum tuberosum L.) is one of the crucial essential staple foods on which humanity 34 relies for survival. More than 150 countries and regions worldwide cultivate potato, which serves 35 food, vegetable, and feed functions and has excellent potential for further processing, playing an essential role in ensuring food security ¹⁻³. However, plant pests and diseases can lead to crop failure, 36 seriously restricting agricultural production ⁴⁻⁶. A notable instance of damaging pests encompasses 37 38 Potato Virus Y (PVY), a linear RNA-based pathogen categorized within the Potyviridae family. 39 Since it was initially reported in 1931, PVY had caused significant damage to crop yields by 40 infecting various cash crops from the Solanaceae family, tobacco (Nicotiana tabacum L.) and potato being a prominent example.^{7–9}. Currently, managing such viral pests by agrochemicals serves as the 41 42 most direct route, but there were yet satisfactory results from the exploration of commercial drugs 43 like Ningnanmycin (NNM), ribavirin, and moroxydine for the control of PVY-induced disease (Figure 1A), which in turn sparked considerable interest in the development of new molecular 44 entities for PVY management¹⁰⁻¹². 45

The coat protein (CP) plays multiple roles in the PVY life cycle, including virus particle 46 47 assembly, virus movement, and aphid transmission, etc. ¹³. Consequently, CP is considered a 48 promising candidate for biochemical targeting in developing new pesticides with antiviral properties that bind to CP, thereby suppressing virus infection ¹⁴. Indeed, CP has emerged as an ideal drug 49 50 target and has been widely utilized in developing antiviral agents against plant viruses. Many CP inhibitors with diverse chemical structures and activities against various plant viruses have been 51 reported (Figure 1B)¹⁵⁻¹⁸. Despite identifying critical sites on the CP that significantly mediate 52 viral pathogenicity, a deeper understanding of molecular interactions remains lacking. Recently, we 53 54 confirmed that chiral arylimidazole-fused compound A could competitively bind to the R¹⁹¹ site of PVY CP, interfering with the mutual effect between CP and the host factor ntCPIP at the molecular 55 level, thereby preventing virions from moving across plant cells (Figure 1C)¹⁵. Therefore, it is 56 57 essential to identify additional key sites that significantly influence viral behavior and elucidate the 58 molecular mechanisms of drugs, providing a basis for developing effective and sustainable pest 59 management strategies.

60 In 2019, Podobnik and co-workers reported the structural basis of PVY CP, which 61 demonstrated that the interaction between viral RNA (vRNA) and CP is important for virion to maintain a stable helical configuration ¹³. This pioneering work provides opportunities for the design 62 of antiviral agents based on PVY CP. It has been suggested that the conserved residues Ser¹²⁵, Arg¹⁵⁷, 63 and Asp²⁰¹ of PVY CP are related to the binding of vRNA. Hence, using drugs to interfere with 64 these sites may hinder the interaction between CP and vRNA, thereby affecting downstream 65 66 pathogenic behaviors. Moreover, because these sites belong to conserved regions, this enables the 67 broad spectrum of the drug and makes it less likely to develop drug resistance. Unfortunately, no 68 molecules have yet been discovered to enter this cavity and bind to its sites.

Here, we disclose a case of phytovirucide that could form a stable H-bond with Ser¹²⁵ of PVY CP to exert antiviral properties (**Figure 1D**). The present investigation explores a range of innovative compound derivatives of 2,2-dimethyl-2*H*-chromene that were designed by using a molecular hybridization strategy. Notably, these target compounds bear a unique Schiff base unit and a phenolic hydroxyl group, which introduces additional hydrogen bond donors. This inclusion facilitates the formation of intramolecular hydrogen bonds with nitrogen atoms in neighboring Schiff base units, creating a six-membered ring and enhancing structural stability, as confirmed by

X-ray diffraction results (CCDC No. 2297246). Many of our synthesized derivatives displayed 76 77 potent curative, protective, and inactivating effects against PVY during bioactivity assays. Among 78 them, compound C50, identified through constructing a three-dimensional quantitative structure-79 activity relationship (3D-QSAR) model, demonstrated significantly better inactivation effects (EC₅₀ 80 = 53.3 μ g/mL) compared to the commercial drug NNM (EC₅₀ = 73.7 μ g/mL). Combining molecular 81 docking, molecular dynamics simulation, and a range of biological experiments allowed us to discover that compound C50 can specifically bind to the conserved residue Ser^{125} on CP. We 82 83 confirmed that this site is crucial for the interaction between vRNA and CP, which can be 84 competitively inhibited by our drug, leading to dysfunction in the assembly of virus particles. Our 85 research enhances knowledge of the mode of action of PVY-targeting drugs and lays the groundwork 86 for designing innovative inhibitors of virus CP assembly at the molecular level.

87

88 RESULTS

89 Chemistry

Firstly, p-phenylenediamine and substituted sulfonyl chloride were reacted in dichloromethane 90 91 solvent, utilizing triethylamine as an acid scavenger, to synthesize intermediates designated as 92 A1-A41; secondly, 2,4-dihydroxybenzaldehyde was used as raw materials with 3-methyl-2-butena 93 or citral, and refluxed in an ethanol solution of triethylamine and anhydrous calcium chloride, and 94 carried out the ring-combining reaction to acquire intermediates B1 and B2. Finally, A1-A41 and 95 B1 and B2 were used as raw materials and heated to reflux with ethanol as a solvent to obtain 2,2-96 dimethyl-2H-chromene derivatives C1-C50 with sulfonamide units (Figure 2A). As X-ray 97 diffraction results showed, the hydroxyl group at the 5-position on the chromene ring of compound 98 C6 formed an intramolecular hydrogen bonding interaction with the nitrogen atom on the imine unit 99 at the 6-position. This force formed a six-membered ring structure, which led to a more stable 100 structure of the synthesized title compounds of the salicylaldehyde Schiff base class (Figure 2B). It 101 is clear that the chemical properties of this class of compounds fully exhibit the *E* configuration.

102

103 Assay anti-PVY activity

104 Utilizing the half-leaf blight method with Chenopodium amaranticolor as the indicator host and 105 NNM as the positive control, the anti-PVY efficacy of all target compounds at a concentration of 106 500 μ g/mL was assessed and tabulated in **Table 1**. The target compounds exhibited notable to 107 superior antiviral potency towards PVY, with 13 of the target compounds (C7, C10, C16, C24, C26, 108 C27, C28, C29, C32, C34, C36, C37, C38, and C50) having curative activities and protective 109 activities in the range of 60-70%, significantly higher than those of the commercial agent NNM (50.1%, 50.3%). Compounds C6, C26, C27, C32, C40, and C50 showed comparable or better 110 111 activity than NNM for inactivating effect. Based on the excellent inactivating performance of this 112 group of compounds (inhibition rate around 80%), we determined the EC₅₀ values of inactivating 113 activity for all target compounds. We found that target compounds C6 and C50 showed better 114 inactivating EC₅₀ values of 56.7 µg/mL and 53.3 µg/mL compared to the control agent NNM (73.7 μ g/mL). The inactivating property of compound C50 at different concentrations is shown in Figure 115 116 **3A.** NNM serves as the positive control, whereas the negative control is positioned on the left flank 117 of the leaf. Given the excellent inactivating activity of C6 and C50, we next studied the preliminary 118 mechanism behind this.

120 Compound C50 inhibits systemic infection of PVY

121 To evaluate the effect of C50 on PVY infection, we infiltrated the Nicotiana benthamiana 122 plants with Agrobacterium carrying the PVY infectious clone pCamPVY-GFP. As depicted in Figure 3B, a notable presence of strong green fluorescence was observed in the stems and apical 123 124 shoot leaves of the control group (DMSO, 500 μ g/mL). On the other hand, there was a noteworthy 125 decrease in green fluorescence observed in both the C50 (500 μ g/mL) and NNM (500 μ g/mL) 126 treatment cohorts, with a more substantial reduction observed in the C50 treatment cohort. Reverse 127 transcription quantitative real-time polymerase chain reaction (RT-qPCR) analysis revealed that, as opposed to the control group (in which PVY RNA accumulation was normalized as 100%), the PVY 128 129 RNA accumulation level in C50 and NNM- treated N. benthamiana plant was significantly reduced 130 to 18.9% and 51.8%, respectively, in the systemic leaves of the infected N. benthamiana leaves 131 (Figure 3C). The RT-qPCR results for GFP were consistent with that of PVY RNA (Figure 3D). 132 These results showed that C50 is a potential antiviral agent against PVY.

133

134 **3D-QSAR studies**

135 Using Compound C6 as a template, we aligned compounds in the training set and developed 136 CoMFA and CoMSIA models incorporating the anti-PVY inactivity of the targets. The parameters 137 of these models were then scrutinized to assess their reliability. Table 2 displays the q^2 , r^2 , SEE, and F values of the built CoMFA and CoMSIA models. Both models' q^2 , r^2 , and F values exhibited more 138 139 significance than 0.5, 0.8, and 100, respectively. Additionally, the SEE values were comparatively 140 less, suggesting that the built models possess predictive capabilities. According to the data in 141 Figures 4A & 4B and Table S1, the experimental and theoretical pEC₅₀ values closely matched. 142 The residuals in the analysis had absolute values below 0.3. The linear correlation coefficients for 143 experimental vs predicted pEC₅₀ were high, at 0.9339 for the CoMFA and 0.9196 for the CoMSIA. 144 These findings provide further evidence that both CoMFA and CoMSIA models effectively forecast 145 the anti-PVY activity of the compounds.

146 The CoMFA model showed nearly equal contributions from steric (50.5%) and electrostatic 147 (49.5%) fields, emphasizing their joint importance for anti-PVY activity. Both factors should be 148 addressed when optimizing compounds. Based on the CoMFA model, it can be seen that the position 149 of R^1 is mainly a green module (Figure 4C), indicating that increasing the spatial site resistance of R^{1} is advantageous to the anti-PVY inactivating activity, in which substitution of the aromatic ring 150 with a significant site resistance is better than that of the aliphatic or cycloalkanes with a small site 151 resistance, such as C7 ($R^1 = 3$ -FC₆H₄, $R^2 = H$, EC₅₀ = 100.3 $\mu g/mL$) < C29 ($R^1 = (CH_2)_3 CH_3$, $R^2 =$ 152 H, $EC_{50} = 682.1 \,\mu g/mL$). Positions are mainly yellow modules, indicating that increasing the spatial 153 site resistance of R^2 is detrimental to anti-PVY inactivating activity, such as C41 ($R^1 = 4$ -ClC₆H₄, 154 $R^2 = CH_2CHC(CH_3)_2$, $EC_{50} = 780.8 \ \mu g/mL) > C6 \ (R^1 = 4-ClC_6H_4, R^2 = H, EC_{50} = 56.7 \ \mu g/mL)$, 155 C42 ($R^1 = 4$ -CH₃C₆H₄, $R^2 = CH_2CHC(CH_3)_2$, EC₅₀ = 503.2 μ g/mL) > C50 ($R^1 = 4$ -CH₃C₆H₄, $R^2 = 6$ 156 H, EC₅₀ = 53.3 μ g/mL), C48 (R¹ = 3,5-di-FC₆H₃, R² = CH₂CHC(CH₃)₂, EC₅₀ = 716.6 μ g/mL) > C26 157 $(R^1 = 3,5-di-FC_6H_3, R^2 = H, EC_{50} = 100.3 \mu g/mL)$. In CoMFA's electrostatic field (Figure 4D), R^{1} 's 158 dominant red module hints at enhanced anti-PVY activity with electron-withdrawing R¹ groups. 159 The position of R² is mainly a blue module, indicating that introducing electron-donating groups in 160 161 R^2 can be conducive to raising the anti-PVY inactivating activity. In summary, the R^1 substituent is 162 an electron-withdrawing group, and the introduction of a large-site-resistance substituent is 163 beneficial to increase the activity; the reduction of the spatial site resistance of the 2-position methyl

group (R^2) of the pyran ring and incorporating an electron-donating group boosts anti-PVY activity. 164 165 Figures 4E & 4F depict the three-dimensional and electrostatic fields belonging to the CoMSIA model. Assessment of color block significance and roles yields conclusions mirroring those of the 166 167 CoMFA model. The hydrophobic field of the CoMSIA model is displayed in Figure 4G. A yellow 168 region surrounds R¹, indicating that introducing hydrophobic groups promotes the anti-PVY activity 169 of target compounds. In contrast, the white color does the opposite. Figure 4H shows the hydrogen bonding receptor field of the CoMSIA model, with a larger red module at the position of the 170 171 sulfonamide group oxygen atom, suggesting that the hydrogen bonding receptor provided by the 172 sulfonamide double-bonded oxygen favors the anti-PVY activity of the compounds.

The analysis conducted using the model above demonstrated that incorporating a sizeable electron-withdrawing group at R^1 and a small electron-donating group at R^2 resulted in a favorable enhancement in the anti-PVY activity belonging to compounds under investigation. Furthermore, it was observed that the presence of a sulfonamide group played a crucial role as a hydrogen-bonding acceptor group (**Figure 4I**).

178

179 Molecular docking and molecular dynamics simulation

180 To verify whether the compounds have interaction forces with PVY CP and to find their binding 181 sites, compounds C6 and C50 were molecularly docked to PVY CP, respectively, employing ultraprecise XP docking under the Schrödinger software module. Detailed examination of XP docking 182 and MM-GBSA outcomes revealed that detailed examination of XP docking and MM-GBSA 183 184 outcomes revealed compounds C6 and C50 had docking scores of -6.045 and -5.999 with PVY CP, 185 and MM-GBSA results were -34.15 kcal/mol and -34.44 kcal/mol, with low docking scores and free energies of binding, which indicated that C6 and C50 were stable in binding to PVY CP. The 186 187 molecular docking results showed that C6 penetrated deep into the interior of the active pocket of PVY CP (Figure S1), developing one hydrogen bond with residue Ser¹²⁵ of PVY CP with a bond 188 length of 1.91 Å, two hydrogen bonds, and one salt bridge with residue Asp²⁰¹ with bond lengths of 189 2.43 Å, 2.34 Å, and 4.70 Å, and one salt bridge and two π -Cation bonds with residue Arg¹⁸⁸ with 190 191 bond lengths of 4.02 Å, 3.70 Å, and 4.63 Å. Compounds C50 and C6 are structurally distinguished only by methyl and chlorine atoms. The docking results are similar to the C6 results, showcasing 192 C50's deep penetration into PVY CP's active site. It forms a hydrogen bond with Ser¹²⁵ (2.00 Å), 193 194 two with Asp²⁰¹, and a salt bridge (4.75 Å, 2.42 Å, 2.38 Å). Additionally, it establishes a salt bridge 195 and π -Cation bond with Arg¹⁸⁸, measuring 3.38 Å and 4.58 Å, respectively (Figure 5A). In summary, it can be seen that the compounds C6 and C50, which are highly inactivating activity compounds 196 for PVY, both form strong hydrogen bonding interactions forming hydrogen bonds with Ser¹²⁵ of 197 198 PVY CP at lengths of 1.91 Å and 2.00 Å.

199 Molecular dynamics trajectory simulations of C50 with PVY CP were carried out for 100 ns, 200 and then molecular dynamics trajectories were subsequently scrutinized. The conformational 201 stability of RMSD against simulation time is shown in the figure, where smaller fluctuations indicate 202 that stable conformations were obtained for all the complexes. The analysis reveals a stable C50-PVY CP complex post-15 ns, indicating system equilibrium (Figure 5B). RMSF assesses local 203 204 protein chain variations, highlighting peaks as areas experiencing the most significant fluctuations 205 in the simulation. As shown in Figure 5C, the results showed that the protein displayed significant structural pliability within the residue region 12-35AA and 190-225AA after the binding of C50 to 206 207 PVY CP.

Throughout the simulation, protein-ligand engagements are tracked and classified into four primary types: H-bond, hydrophobic forces, ionic linkages, and aqueous bridges. **Figure 5D** illustrates the pivotal amino acids crucial for **C50** binding to PVY CP protein are Ser¹²⁵, Asn¹²⁷, Asp²⁰¹, Ala²²⁴, Phe²³³, and Arg²⁴⁸. **Figure 5E** demonstrates the interactions between the specific amino acids of **C50** and PVY CP in terms of time during the entire trajectory. The outcomes indicate that amino acid residues Ser¹²⁵, Asp²⁰¹, Ala²²⁴, Phe²³³, Thr²⁴¹, Arg²⁴⁸, and Glu²⁵² have multiple contacts with the ligand.

The ligand torsion diagram outlines the transformation of ligand conformations across rotatable bonds during the simulation path. The upper portion of **Figure 5F** depicts a 2D illustration of a ligand, featuring color-coded rotatable bonds. Each bond's torsion is visually represented by a matching colored dial plot and bar graph. The dial plot in the bottom half of **Figure 5F** illustrates the torsion's conformational changes during the simulation. The bar graph condenses the dial plot data, displaying the probability density distribution of the torsion.

The above results show that compound **C50** penetrated deep into the conserved folding region of PVY CP and interacted with amino acid residues such as Ser¹²⁵ and Asp²⁰¹. A strong hydrogen bonding interaction was formed with Ser¹²⁵ suggesting that Ser¹²⁵ may be the key site for binding 224 *2H*-chromene derivatives containing sulfonamide structure to PVY CP.

225

Purification of the wild-type protein PVY CP^{wt} and the mutant protein PVY CP^{S125A} and microscale thermophoresis assay

228 PVY CP is involved in various biological functions and is pivotal in the life cycle of PVY. Hence, it is regarded as a promising candidate protein for antiphytoviral agent screening. The PVY 229 CP^{wt} encoding sequence was constructed into the expression vector pET-32a (+) using whole gene 230 231 synthesis methodology. The resultant plasmid was named pET-PVY CPwt. The codon at position 232 125 was mutated to Alanine based on the pET-PVY CPwt. The resultant plasmid was named pET-PVY CP^{S125A}. The PVY CP^{wt} and its mutant CP^{S125A} were expressed and purified according to the 233 234 previous protocol. SDS-PAGE results showed that an expected band of about 50 kDa was observed. 235 After enterokinase digestion and His-tag purification resin, a specific band of about 33 kDa was 236 observed (Figure S2). This result is consistent with the theoretical value ¹³.

237 Microscale thermophoresis (MST) serves as a potent technique for quantifying the strength of interactions occurring between small molecules and their respective ligands ^{19, 20}. The binding 238 strength between specific target drugs and PVY CPwt was investigated in vitro using MST, as 239 240 depicted in Figure 5G and Figure S3. This experimental approach aimed to acquire insights into 241 how target compounds exert their influences against PVY. The findings indicated that the binding strength of target compounds to PVY CP^{wt} was in line with their effectiveness in anti-PVY. The K_d 242 243 values for binding of compounds C6 and C50 to PVY CP^{wt} were 4.4 μ M and 1.6 μ M, respectively, 244 indicating that compounds C6 and C50 have a strong affinity for PVY CP, which is superior to NNM ($K_d = 8.6 \mu M$). Compound C42 had low inactivating activity against PVY (EC₅₀ = 503.2 245 μ g/mL). Its binding to PVY CP^{wt} was also relatively weak ($K_d = 497.1 \ \mu$ M). To further confirm that 246 the serine at position 125 of the PVY CP protein is a key site of action for the binding of PVY by 247 248 2,2-dimethyl-2H-chromene derivatives containing a sulfonamide unit, the serine at position 125 of the PVY CP protein was fixed point mutated to an alanine called PVY CP^{S125A}. Similarly, the affinity 249 of target compounds for interacting with PVY CP^{S125A} was tested. The findings indicated a markedly 250 diminished binding affinity between the target compounds and PVY CP^{S125A} , such as the K_d value 251

of compound **C50** was decreased from $1.6 \,\mu$ M to $125.4 \,\mu$ M. Among them, the weakening of binding between NNM and PVY CP^{S125A} was not very obvious, indicating that the serine at position 125 of PVY CP protein is not a key site for binding of NNM to PVY CP, which further indicates that this site is the critical site for anti-PVY of 2,2-dimethyl-2*H*-chromene derivatives containing sulfonamide unit.

257

258 Effects of mutation at position Ser¹²⁵ of CP on PVY accumulation levels

To further validate the possibility of the residue Ser¹²⁵ of CP function as a potential target site 259 for the small molecule C50 in vivo, we mutated the codon at position 125 in CP based on the 260 261 infectious clone pCamPVY-GFP using site-directed mutagenesis, the resultant plasmid was named pCamPVY^{S125A}-GFP (the produced virus named as PVY^{S125A}-GFP) (Figure 6A). The plasmids 262 pCamPVY-GFP and pCamPVY^{S125A}-GFP were transformed into Agrobacterium GV3101 by freeze-263 thawing, respectively. Agrobacterium carrying those plasmids were infiltrated on the fully expanded 264 265 leaves of four to six-week-old N. benthamiana plants. At 7 days post-agroinfiltration (dpai), a strong green fluorescence was observed on the systemic leaves of N. benthamiana plants infiltrated with 266 Agrobacterium carrying pCamPVY-GFP under UV light. However, no green fluorescence was 267 268 observed on the N. benthamiana plants infiltrated with Agrobacterium cells carrying pCamPVY^{S125A}-GFP (Figure 6B). Western blot and RT-qPCR analyses revealed a substantial 269 accumulation of PVY CP in the systemic leaves of wild-type PVY GFP-infected N. benthamiana 270 271 plants. In contrast, only minimal traces were detected in the corresponding leaves of PVY^{S125A}-GFPinfected N. benthamiana (Figures 6C & 6D). In summary, PVY CP^{S125} significantly affected virus 272 273 accumulation in N. benthamiana.

Virus particle assembly and movement are two critical processes for potyviruses infection. To 274 275 test the effect of mutation on the Ser¹²⁵ in CP on PVY cell-to-cell movement, we adjusted the OD₆₀₀ value of Agrobacterium cells carrying pCamPVY-GFP and pCamPVY^{S125A}-GFP to 0.5, and then 276 277 further diluted 1000 times. The diluted Agrobacterium cells was infiltrated on four-to-six-week-old 278 N.benthamiana plants. The cell-to-cell movement was observed using a laser confocal microscope 279 at 3 dpai. Confocal microscopy analysis showed that PVY could move to multiple cells in the N. 280 benthamiana plants inoculated with PVY-GFP. However, the number of cells displaying green fluorescence was much less in the N. benthamiana plants inoculated with PVY^{S125A}-GFP (Figure 281 6E). The findings suggested that the PVY CP^{S125} mutation did not impair the cell-to-cell movement 282 283 of PVY-GFP.

Based on the above experimental results, we found that the substitution of Ser¹²⁵ with Alanine 284 285 in CP significantly affected the PVY accumulation in N. benthamiana plants but did not abolish its cell-to-cell movement. Disturbing the interaction between CP and vRNA reduces the stability of the 286 287 virus particles. Previous studies have shown that Ser¹²⁵ of CP binds viral RNA¹³. It is interesting to know whether the mutation on Ser¹²⁵ affects the virus particle assembly. We purified the virus 288 particles of PVY-GFP and PVY^{S125A}-GFP, respectively, followed by the previously reported 289 protocol ¹⁵. The stability of PVY virus particles is derived from many CP-vRNA interactions, and 290 Ser¹²⁵ serves as one of the sites where the PVY CP binds to the vRNA. Mutation of this polar amino 291 acid residue (Ser) to the non-polar amino acid Ala may impact virus particle assembly. Therefore, 292 we further observed the effect of PVY CP^{S125} on virus particle assembly by transmission electron 293 microscopy. Wild-type PVY-GFP and mutant PVY CP^{S125A}-GFP were infiltrated into leaves of age-294 295 appropriate N. benthamiana, and virus particles were collected for virus particle extraction at 5 dpai.

The purified specimens were treated with 1% phosphotungstic acid staining, followed by morphological observation of the refined PVY virus particles via transmission electron microscopy. Intact PVY virus particles, curved linear particles of approximately 11 nm \times 680–900 nm in size (**Figure 6G**), were observed in PVY-GFP and PVY-GFP^{S125A}-treated leaves, which is in accordance with literature reports ¹⁵. Notably, the viral particle count in PVY^{S125A}-GFP samples was markedly lower than that in PVY-GFP-infected counterparts (**Figure 6F**). Thus, PVY CP^{S125} is essential for forming PVY virus particles in plants.

303

304 **DISCUSSION**

305 Natural compounds serve as a vital resource in the pursuit of innovative antiviral medications. 306 Among them, 2,2-dimethyl-2H-chromene is present in large quantities in nature and possesses a vast range of antiviral activities. Calanolide A, an extract from the tropical rainforest plant 307 Calophyllum lanigerum, was one of the first natural products to be discovered with anti-HIV-1 308 activity ²¹. Recently, a new indole alkaloid, Isoaspergilline A, has been isolated from the 309 fermentation product of the tobacco endophyte fungus Aspergillus versicolor and and has been 310 311 shown to have excellent anti-TMV activity ²². Sulfonamide derivatives have always been the star 312 structures in drug discovery and development, which increase the water solubility of drugs and provide additional hydrogen-bonded receptors. Previous work by our group has shown that 313 314 sulfonamides have excellent antiphytoviral activity, and some of the compounds can bind specifically to CP^{15, 16, 23-26}. Salicylaldehyde Schiff base derivatives (E) -2,4-dichloro-6- (((3-315 methoxyphenyl) imino) methyl) phenol functions as an synthetic elicitors to enhance the immune 316 response of plants. To resist persecution by harmful organisms, and salicylaldehyde Schiff base 317 318 derivatives also have excellent antiviral activity ^{27, 28}.

319 In drug design, a molecular hybridization strategy is an important approach based on 320 combining two or more molecular fragments with different biological activities into a new 321 molecular entity by chemical or biological means. This strategy aims to combine the individual fragments' strengths to improve the biological activity, pharmacokinetic properties, and selectivity 322 of the new molecule against a specific target ^{29, 30}. Here, we have developed sulfonamide-containing 323 324 2H-chromene derivatives with anti-PVY activity by a clever combination of a sulfonamide structure 325 that increases the water solubility of the drug and provides an additional hydrogen-bonded receptor 326 and a natural product structure 2,2-dimethyl-2H-chromene, which has antiviral activity. Notably, 327 these target compounds carry a unique salicylaldehyde Schiff base unit, as confirmed by X-ray 328 diffraction results, where the phenolic hydroxyl group forms intramolecular hydrogen bonds with 329 nitrogen atoms in the neighboring Schiff base unit to constitute a six-membered ring, which enhances the structural stability of the target compounds. 3D-QSAR plays a crucial role in drug 330 331 design ^{31–33}. This approach employs theoretical modeling and statistical methodologies to 332 investigate the quantitative correlation between the 3D molecular configurations of a compound 333 series and their respective biological outcomes., and it can reveal conformational relationships and 334 guide structural modifications. Once a reliable 3D-QSAR model has been established, it facilitates 335 the prediction of the biological potency of compounds. A 3D-QSAR model was constructed based 336 on the EC₅₀ value of the anti-PVY inactivating activity of this series of compounds, which further 337 guided the synthesis of C50. The bioactivity assessment revealed C50 to exhibit potent anti-PVY 338 efficacy, with an EC₅₀ of 53.3 μ g/mL for its inactivating activity, outperforming the standard agent 339 NNM which had an EC₅₀ of 73.7 μ g/mL. This led to our interest in the mechanism of antiviral action

340 of **C50**.

Currently, there are two primary approaches for tackling plant viral infections. One prominent method is stimulating the host immune system against viral invasion through plant immune inducers 24–26, 32, 34. The other is the use of antiviral drugs, which target the virus's genetic material or essential proteins to stop its pathogenic behavior, a simple mode of action, and a much faster onset of action $1^{4-17, 35}$. The CP of plant viruses holds a pivotal function throughout the viral lifecycle, including assembly and movement of virus particles, transmission by vectors, and RNA translation and replication, and is a key biochemical target for developing new pesticides $1^{4-16, 35, 36}$.

348 Moreover, based on the excellent inactivating activity of C50 against PVY, molecular docking, 349 and molecular dynamics simulations revealed that there is a strong interaction between C50 and the 350 active pocket in the conserved folding region of the PVY CP, where a segment of the amino acid 351 sequences within the specified area are among all the flexible filamentous viruses highly conserved ^{13, 36, 37}. Thus, scientists consider this a key target for developing antiviral agent drugs ³⁷. Therefore, 352 interfering with these sites using drugs may hinder the interaction between CPs and vRNA, thereby 353 354 affecting downstream pathogenic behavior. In addition, since these sites are conserved regions, this allows for a broad range of drug action and less likely resistance. What is exciting is that C50 forms 355 a strong hydrogen bonding interaction with amino acid residue Ser¹²⁵ in the conserved folding 356 region of CP, which serves as one of the key sites for CP binding to vRNA¹³. It is reasonable to 357 believe that our drug can compete with viral ssRNAs to bind to this site, which leads to dysfunction 358 359 of viral particle assembly and thus inhibits viral infestation in plants.

In the early stages of drug discovery, targeted mutagenesis strategies and MST techniques can be used to validate the effectiveness of potential drug targets ^{15, 20}. The mechanism of action and target selectivity of drug molecules can be initially assessed by altering critical residues of the target protein and observing the attachment of the drug molecule to the mutated target protein. The targeted mutation of Ser¹²⁵ on PVY CP to Ala¹²⁵ combined with MST assay revealed that the attachment of **C50** to PVY CP was noticeably diminished, which further indicates that this site is the critical site for anti-PVY of **C50**.

Previous research has demonstrated that the CP of PVY engages with vRNA twice: firstly, at 367 three critical conserved vRNA-binding amino acids, Ser¹²⁵, Arg¹⁵⁷, and Asp²⁰¹, within a structured 368 region; secondly, at a non-conserved site, Ser²⁴⁰, positioned in the C-terminal segment of CPn ¹³. 369 From a functional standpoint, mutations in the conserved Arg and Asp residues of CPs disrupt the 370 in vitro assembly of potyvirus Johnsongrass mosaic virus and hinder the assembly and intercellular 371 trafficking of potexvirus Pepino mosaic virus within plants ^{37, 38}. For PVY, however, there is no 372 information on whether mutations in conserved residues of the CP affect virus function. We 373 successfully expressed and purified a mutant CP (PVY CP^{S125A}) in E. coli that mutated Ser at 374 position 125 in the conserved region of the PVY CP to Ala, and this result preliminarily 375 demonstrated that the Ser¹²⁵ mutation did not affect the in vitro assembly of the CP. We constructed 376 the mutant PVY^{S125A}-GFP using a targeted mutagenesis strategy to verify this target's validity and 377 other functions in vivo. Infestation of N. benthamiana with Agrobacterium tumefaciens containing 378 the PVY^{S125A}-GFP plasmid revealed a considerable decrease in the accumulation of PVY in the 379 380 plant, the same as that found after drug treatment. The accumulation of plant viruses in the host is mainly influenced by the replicative assembly and movement of virus particles ^{39, 41-43}. Our 381 experimental results by confocal microscopy and analysis indicated that the mutation in the PVY 382 383 CP^{S125} locus would not impede the intercellular dissemination of PVY. but results in fewer virus

particles. The likely reason for this is that the Ser¹²⁵ mutation affects viral assembly, resulting in
 fewer viral particles being formed.

386

387 Conclusion

388 CP is an essential component of utmost utility in viral life cycle, as it unlocks a variety of 389 important downstream pathogenic processes. Efforts in developing effective CP inhibitors with 390 unique action mechanisms continue to the present day; the contribution reported here provides a 391 myriad of 2H-chromene-based antiphytovirals that have shown promising abilities against PVY infection. Compound C50 stood out among the 50 products and control commercial drug. In vitro, 392 the binding site on CP was discovered and verified as Ser¹²⁵ using molecular docking paired with 393 MST. Several tests conducted in vivo indicated that Ser¹²⁵ is important during the mutual effect 394 395 between PVY CP and vRNA, which is necessary for the assembly of PVY virions. Compound C50 396 can dramatically decrease the viral pathogenicity through competitively inhibiting position Ser¹²⁵ 397 (Figure 7). In conclusion, this study provides a theoretical basis and potential action sites for the 398 development of CP-based virus assembly inhibitory pesticides, offering new avenues for combating 399 plant viral infections.

400

401 Limitations of the study

402 Although compounds with good inhibitory activity against PVY were identified, subsequent 403 structural optimization of this class of compounds is required to discover more active and less costly 404 lead small molecules against plant viruses. The antiviral efficacy of C50 was initially substantiated 405 in this study solely through molecular docking and dynamic simulations, MST and preliminary 406 investigations on the effect of mutated amino acid sites on the function of PVY virus infestation, 407 movement and viral particle formation were carried out. However, whether the drug can make host 408 plants resistant to the disease is still unknown. In addition, changes in the binding affinity of CP to 409 vRNA after Ser¹²⁵ mutation or drug treatment have not been verified by bio-layer interferometry 410 technology.

411

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420

421 AUTHOR CONTRIBUTIONS

422 Conceptualization, Runjiang Song and Xiong Yang; Methodology, Deguo Liu, Chunle Wei and
423 Yanping Tian; Investigation, Xiong Yang, Deguo Liu, Jianzhuan Li, and Chunni Zhao; Writing –
424 Original Draft, Xiong Yang; Writing – Review & Editing, Runjiang Song and Yanping Tian;
425 Funding Acquisition, Baoan Song and Runjiang Song; Resources, Baoan Song, Xiangdong Li and
426 Yanping Tian; Supervision, Runjiang Song, and Baoan Song.

428 DECLARATION OF INTERESTS

- 429 The authors declare no competing interests.
- 430

431 FIGURE TITLES AND LEGENDS

432 Figure 1. Rational design of 2*H*-chromene-based antiphytovirals.

433

434 Figure 2. Production of the intended compounds.

435 (A) Manufacturing pathways for the target molecules. Reagent and conditions: (*i*) Et₃N, 436 DCM, 0 °C 30 min, then at rt 3 h, 50–80% yield; (*ii*) Et₃N, CaCl₂, EtOH, reflux, 3 h, 50–60% yield; 437 (*iii*) cat. AcOH, MeOH, reflux, 5 h, 40–75% yield. (B) X-ray crystal structure of compound 438 C6 (CCDC: 2297246).

439

440 Figure 3. Effect of C50 against PVY infection.

(A) Effect of C50 against PVY infection in *Chenopodium amaranticolor*. Ningnanmycin (NNM) 441 442 was used as a control. The concentration was marked below the inoculated leaves. (B) Green 443 fluorescence expression map of Nicotiana benthamiana leaf blades treated with PVY-GFP 444 infectious C50 and NNM under UV illumination. (C and D) Quantitative assessment of PVY CPand 445 GFP accumulation in systemically infected leaves of N. benthamiana plants at 7 days postagroinfiltration (dpai), utilizing RT-qPCR method. After injection of Agrobacterium with 446 447 pCamPVY-GFP, a solution of 1% Tween 80 containing C50 (500 µg/mL), NNM (500 µg/mL), and 448 DMSO (as a control) was sprayed on the leaves of N. benthamiana. RT-qPCR normalization was 449 achieved using EF1 α as an internal control. Data are presented as mean \pm SD from three biological 450 replicates per treatment, with statistical significance indicated by different letters (p < 0.05, one-451 way ANOVA).

452

453 Figure 4. 3D-QSAR Analysis.

(A) CoMFA and (B) CoMSIA models for comparing experimental versus predicted pEC₅₀. (C and
D) CoMFA 3D isopotential maps illustrating (C) steric and (D) electrostatic contributions. (E-H)
CoMSIA 3D isopotential maps depicting (E) steric, (F) electrostatic, (G) hydrophobic, and (H)
hydrogen bond acceptor fields. (I) Relationship between structure and anti-PVY activity.

458

459 Figure 5. Ser¹²⁵ is a key target for binding compound C50 to PVY CP.

(A) Computational binding analysis of C50 to PVY CP using molecular docking techniques. (B-F)
Comprehensive results of molecular dynamics simulations of C50 and PVY CP. (B) RMSD analysis,
(C) RMSF analysis, (D and E) Interaction analysis, (F) Ligand torsion diagram. (G) analysis
comparing the binding of PVY CP^{wt} and PVY CP^{S125A} mutant proteins to compounds

464

465 Figure 6. Effect of mutation on Ser¹²⁵ in CP on PVY infection.

(A) Illustrative representation of the pCamPVY-GFP genome structure, highlighting the Ser¹²⁵
residue (marked by red arrows) within the core domain of PVY CP. The blue-bordered box
encapsulates the site-specific mutants, wild-type plasmids, viruses, and their respective sequences.
(B) Comparative visualization of disease symptoms (top) and green fluorescent expression (bottom)
under UV illumination in *N. benthamiana* leaves inoculated with wild-type and mutated PVY strains.
(C and D) Quantitative assessment of PVY CP accumulation in systemically infected leaves of *N*.

benthamiana plants at 7 dpai, utilizing RT-qPCR (C) and Western blot (D) methods. RT-qPCR 472 473 normalization was achieved using EF1 α as an internal control, while Ponceau S-stained RuBisCO 474 served as a loading control. Data are presented as mean \pm SD from three biological replicates per treatment, with statistical significance indicated by different letters (p < 0.05, one-way ANOVA). 475 476 (E) Examination of cell-to-cell movement dynamics in *N. benthamiana* plants infected with wild-477 type and mutated PVY strains at 3 dpai. (F) Enumeration of PVY particles within 70 μ m² 478 microscopic fields, presented as mean \pm SD derived from five fields per treatment. Statistical 479 significance is denoted by different letters (p < 0.05, one-way ANOVA). (G) Particles of PVY-GFP, and PVY^{S125A}-GFP under transmission electron microscope. 480

481

482 Figure 7. Mechanism of anti-PVY action of compound C50.

483

484 **TABLE TITLES AND LEGENDS**

485 Table 1. Antiviral efficacies of target compounds directed against PVY.

Compd.		Curative activity (%) ^a	Protective activity (%) ^a	Inactivating activity (%) ^a	EC50 for inactivating activity (µg/mL)
C1	$R^1 = 4$ -OCH ₃ C ₆ H ₄ , $R^2 = H$	50.7 ± 5.5	62.2 ± 6.1	55.1 ± 6.0	441.2 ± 5.2
C2	$R^1 = 4$ - $CF_3C_6H_4$, $R^2 = H$	63.1 ± 5.8	58.9 ± 3.5	71.5 ± 3.5	$255.7\pm.7.2$
C3	$R^1 = (CH_2)_7 CH_3, R^2 = H$	12.3 ± 2.1	39.4 ± 3.7	40.1 ± 3.0	605.1 ± 12.2
C4	R^1 = Thiophene-2-yl, R^2 = H	28.1 ± 1.7	29.3 ± 2.9	38.1 ± 3.1	667.3 ± 20.1
C5	$R^1 = 4$ -FC ₆ H ₄ , $R^2 = H$	70.3 ± 5.8	63.0 ± 1.6	66.7 ± 6.4	165.3 ± 8.7
C6	$R^1 = 4$ -ClC ₆ H ₄ , $R^2 = H$	63.1 ± 2.1	57.3 ± 3.5	80.0 ± 1.9	56.7 ± 4.2
C7	$R^1 = 3$ -FC ₆ H ₄ , $R^2 = H$	73.7 ± 2.6	70.2 ± 2.6	74.9 ± 2.4	100.3 ± 5.7
C8	$R^1 = 2$ -FC ₆ H ₄ , $R^2 = H$	67.7 ± 2.0	58.4 ± 3.7	58.9 ± 2.6	369.2 ± 9.9
C9	$\mathbf{R}_1 = \mathbf{Pyridin} - 3 - \mathbf{yl}$, $\mathbf{R}^2 = \mathbf{H}$	30.4 ± 2.9	32.7 ± 3.5	60.1 ± 2.7	310.2 ± 8.6
C10	$R^1 = 5$ -Br-Thiophene-2-yl, $R^2 = H$	68.1 ± 3.0	63.3 ± 4.4	50.2 ± 2.4	492.3 ± 13.3
C11	$R^1 = 3$ -CNC ₆ H ₄ , $R^2 = H$	29.1 ± 6.1	26.7 ± 3.7	44.2 ± 4.6	703.3 ± 25.3
C12	$R^1 = 4-NO_2C_6H_4, R^2 = H$	31.8 ± 3.9	31.9 ± 3.9	58.9 ± 4.9	400.3 ± 7.5
C13	$R^1 = Cyclopropane, R^2 = H$	33.4 ± 3.5	27.8 ± 3.5	49.1 ± 3.1	531.2 ± 20.7
C14	$R^1 = 2,5$ -di-ClC ₆ H ₃ , $R^2 = H$	31.2 ± 4.1	28.6 ± 2.4	35.2 ± 2.8	751.1 ± 19.4
C15	$R^1 = 2,5$ -di-FC ₆ H ₃ , $R^2 = H$	54.1 ± 2.8	54.6 ± 3.1	30.7 ± 2.7	903.2 ± 14.3
C16	$R^1 = 3$ -BrC ₆ H ₄ , $R^2 = H$	61.2 ± 2.8	63.4 ± 3.1	41.1 ± 1.5	688.4 ± 17.2
C17	$R^1 = C_6 H_5, R^2 = H$	56.8 ± 5.6	61.2 ± 2.9	74.3 ± 1.8	168.4 ± 4.8
C18	$R^1 = 4$ - $CH_3CH_2C_6H_4$, $R^2 = H$	57.4 ± 4.3	59.0 ± 2.9	79.1 ± 1.2	202.1 ± 3.8
C19	$R^1 = 2,4,6$ -tri-CH ₃ -C ₆ H ₂ , $R^2 = H$	25.8 ± 2.6	58.7 ± 3.5	56.7 ± 1.9	492.3 ± 11.2
C20	$R^1 = 4$ -tert-butyl C_6H_4 , $R^2 = H$	32.4 ± 5.8	24.5 ± 2.2	20.1 ± 2.5	1003.4 ± 18.2
C21	$R^1 = 4\text{-}CH_3CONHC_6H_4, R^2 = H$	25.7 ± 2.6	22.7 ± 2.9	33.5 ± 2.1	1021.3 ± 20.7
C22	$R^1 = 2$ -CF ₃ C ₆ H ₄ , $R^2 = H$	50.2 ± 2.9	53.7 ± 2.0	45.7 ± 2.0	587.0 ± 9.9
C23	$R^1 = 3,5$ -di-ClC ₆ H ₃ , $R^2 = H$	19.1 ± 3.2	28.0 ± 2.7	23.4 ± 1.4	870.9 ± 17.3
C24	$R^1 = 5$ -Cl-Thiophene-2-yl, $R^2 = H$	67.6 ± 3.1	65.8 ± 2.9	34.5 ± 2.3	753.6 ± 10.3
C25	$R^1 = 2$ -ClC ₆ H ₄ , $R^2 = H$	56.8 ± 5.0	70.9 ± 4.0	67.3 ± 2.7	243.9 ± 10.2
C26	$R^1 = 3,5$ -di-FC ₆ H ₃ , $R^2 = H$	65.2 ± 2.0	63.7 ± 2.9	81.3 ± 3.8	100.3 ± 4.5
C27	$R^1 = 3$ - $CH_3C_6H_4$, $R^2 = H$	63.5 ± 3.0	68.7 ± 3.5	81.8 ± 3.3	90.3 ± 7.1
C28	$R^1 = Benzyl, R^2 = H$	64.4 ± 3.3	68.9 ± 2.3	47.5 ± 1.2	503.1 ± 12.8
C29	$R^1 = (CH_2)_3 CH_3, R^2 = H$	64.2 ± 2.3	66.6 ± 4.5	35.9 ± 5.0	682.1 ± 10.0

Compd.	R^{1}_{O} N H_{O} R^{2}	Curative activity (%) ^a	Protective activity (%) ^a	Inactivating activity (%) ^a	EC ₅₀ for inactivating activity (µg/mL)
C30	$R^1 = 4$ -OCF ₃ C ₆ H ₄ , $R^2 = H$	31.0 ± 2.6	39.6 ± 6.2	32.2 ± 4.7	834.1 ± 18.3
C31	$R^1 = 2,4$ -di-ClC ₆ H ₃ , $R^2 = H$	29.1 ± 2.4	39.7 ± 6.2	35.6 ± 4.7	975.7 ± 20.0
C32	$R^1 = 2,6$ -di-FC ₆ H ₃ , $R^2 = H$	66.8 ± 2.2	63.4 ± 6.0	80.9 ± 3.6	88.3 ± 4.0
C33	$R^1 = 2$ -BrC ₆ H ₄ , $R^2 = H$	52.1 ± 0.6	34.6 ± 3.6	78.1 ± 5.1	139.2 ± 3.2
C34	$R^1 = 4$ -BrC ₆ H ₄ , $R^2 = H$	66.4 ± 2.9	65.2 ± 5.3	79.5 ± 3.9	147.3 ± 8.3
C35	$R^1 = 3\text{-}CF_3C_6H_4, R^2 = H$	48.6 ± 2.6	64.3 ± 4.2	29.2 ± 3.3	803.0 ± 12.3
C36	$R^1 = 3$ -ClC ₆ H ₄ , $R^2 = H$	68.5 ± 4.9	61.0 ± 2.3	44.4 ± 2.0	557.2 ± 14.3
C37	$R^1 = 2$ -F-3-ClC ₆ H ₃ , $R^2 = H$	63.1 ± 3.5	60.2 ± 2.3	59.5 ± 4.2	306.3 ± 11.0
C38	$R^1 = 2$ -CH ₃ -3-ClC ₆ H ₃ , $R^2 = H$	65.5 ± 3.0	61.1 ± 3.1	64.0 ± 3.3	410.3 ± 9.7
C39	$R^1 = 2,5$ -di-CH ₃ -Isoxazol-4-yl, $R^2 = H$	41.0 ± 2.7	47.8 ± 4.4	63.1 ± 3.1	397.1 ± 2.3
C40	$R^1 = L(-)-10$ -Camphoryl, $R^2 = H$	60.2 ± 3.8	53.2 ± 1.8	81.2 ± 1.8	204.3 ± 7.1
C41	$R^1 = 4$ -ClC ₆ H ₄ , $R^2 = \frac{3}{2}$	34.2 ± 4.7	36.9 ± 2.2	36.5 ± 1.5	780.8 ± 15.3
C42	$R^1 = 4$ -CH ₃ C ₆ H ₄ , $R^2 = \frac{3}{2}$	38.8 ± 3.3	44.5 ± 3.4	52.2 ± 3.5	503.2 ± 17.0
C43	$R^1 = 5$ -Cl-Thiophene-2-yl, $R^2 = \frac{\sqrt{-1}}{2}$	33.1 ± 3.7	40.1 ± 2.8	39.7 ± 3.8	512.3 ± 11.0
C44	$R^1 = 3$ -CH ₃ C ₆ H ₄ , $R^2 = \frac{3}{2}$	29.7 ± 6.3	40.0 ± 1.7	34.0 ± 2.8	881.7 ± 17.3
C45	$R^1 = 3$ -F C ₆ H ₄ , $R^2 = \frac{3}{2}$	32.1 ± 1.4	42.4 ± 2.4	32.7 ± 3.3	515.9 ± 9.0
C46	$R^1 = 4$ -CF ₃ C ₆ H ₄ , $R^2 = \frac{3}{2}$	33.5 ± 2.1	41.1 ± 3.8	49.4 ± 3.1	522.4 ± 8.1
C47	$R^1 = 2,6$ -di-FC ₆ H ₃ , $R^2 = \frac{2}{3}$	30.7 ± 4.9	37.1 ± 2.7	44.7 ± 1.2	629.5 ± 11.2
C48	$R^1 = 3,5$ -di-FC ₆ H ₃ , $R^2 = 3^{-1}$	34.7 ± 1.1	31.9 ± 2.7	23.3 ± 2.2	716.6 ± 17.8
C49	$R^1 = 5$ -Br-Thiophene-2-yl, $R^2 = \sqrt[3]{4}$	36.2 ± 1.1	39.4 ± 2.6	40.4 ± 3.3	552.7 ± 9.3
C50	$R^1 = 4$ -CH ₃ C ₆ H ₄ , $R^2 = H$	68.7 ± 3.8	70.1 ± 1.7	84.1 ± 1.0	53.3 ± 4.3
NNM ^b	1	50.1 ± 2.9	50.7 ± 1.8	82.5 ± 1.3	75.7 ± 4.0

486

487

^{*a*} Average of three replicates. ^{*b*} Ningnanmycin (NNM) used as positive control.

488 Table 2. Statistical outcomes belong to the CoMFA and CoMSIA models.

Statistical parameter	CoMFA	CoMSIA	Validation criterion
q^2	0.602	0.619	>0.5
ONC	5	4	
r^2	0.984	0.936	>0.8
SEE	0.042	0.081	
F	408.575	127.810	
	Fraction of fie	eld contributions	
steric	0.505	0.106	
electrostatic	0.495	0.164	
hydrophobic		0.221	
hydrogen-bond donor		0.349	
hydrogen-bond acceptor		0.160	

489

490 STAR METHODS

491 **Resource availability**

492 Lead contact

- Further information and requests for resources and reagents should be directed to and will be fulfilled by the lead contact, Prof. Runjiang Song (songrj@gzu.edu.cn).
- 495

499

501

496 Materials availability

497 Plasmids generated in this study are available from the lead contact, Prof. Runjiang Song498 (songrj@gzu.edu.cn).

500 Data and code availability

All requested data is shared promptly upon receipt of a request.

502 The current study does not encompass the development or reporting of novel coding 503 methodologies.

Additional details necessary for re-examination of the data presented in this article can be sourced from the lead contact upon submission of a request.

506

507 **METHOD DETAILS**

508 Materials and instruments

2,4-dihydroxybenzaldehyde, 3-methylbut-2-enal, citral, *p*-phenylenediamine, and substituted
sulfonyl chloride were purchased from Energy Chemical (Anqing, Anhui, China). The solvents
employed as reagents were of a grade suitable for direct use, without the necessity for any additional
purification steps. A greenhouse was used to cultivate *Nicotiana tabacum* cv. K326, *Nicotiana benthamiana*, and *Chenopodium amaranticolor* plants maintained at a 6/18-h (dark/light)
photoperiod and 25 °C. PVY were propagated in *N. tabacum* cv. K326.

515 A micro melting point apparatus (XGE X4B) was utilized to ascertain the melting point of the 516 compounds. The ¹H, ¹³C, and ¹⁹F spectra were acquired on Bruker AG 400 and a 500 MHz NMR apparatus, respectively, employing TMS as the internal reference and DMSO- d_6 as the solvent 517 518 medium. High-resolution mass spectrometry (HRMS), facilitated by the Thermo Scientific Q 519 Exactive instrument (Waltham, MA, U.S.A.), was employed to determine the molecular weights of 520 the compounds. Furthermore, X-ray crystallographic data were collected using a Bruker Smart Apex 521 CCD area detector diffractometer (Bruker, Germany), which utilized Mo $K\alpha$ radiation for precise 522 measurements.

523

524 **Preparation of compounds C1–C50**

525 The synthesis of intermediate compounds A1-A41 was achieved by adapting the methodology documented in literature source ⁴³. In a 100 mL round-bottom flask, *p*-phenylenediamine (3 mmol) 526 was dissolved in 20 mL of dichloromethane while being maintained under an ice-bath condition. 527 528 Triethylamine (4.5 mmol), serving as an acid scavenger, was subsequently added. A solution of 529 substituted sulfonyl chloride (3.00 mmol) in 5 mL of dichloromethane was then slowly dripped into 530 the mixture. Following the completion of the addition, the ice bath was removed, and the reaction 531 was allowed to proceed with stirring for 3 h. The resulting mixture was concentrated and 532 subsequently purified via silica gel column chromatography, employing a mixture of petroleum 533 ether and ethyl acetate (3:1 ratio) as the eluent. This purification process yielded the intermediate 534 compounds A1-A41.

535 Intermediate compounds **B1** and **B2** were prepared by adapting and partially refining a 536 literature-based procedure ⁴⁴. In this synthesis, 3-methyl-2-butenal or citral (103.6 mmol), CaCl₂

(43.0 mmol), and triethylamine (172.1 mmol) were dissolved in 200 mL of absolute ethanol. 537 538 Subsequently, 2,4-dihydroxybenzaldehyde (51.8 mmol) was introduced into the solution. 539 Formaldehyde (51.8 mmol) was then added, and the reaction mixture was subjected to reflux for 3 540 h. Following this, the ethanol was removed via vacuum distillation. To the residue, 100 mL of 541 distilled water was added, and the pH was adjusted to 5 using dilute HCl. The aqueous mixture was 542 then extracted with ethyl acetate (three times with 100 mL each). The combined organic extracts were washed with saturated brine, dried over anhydrous MgSO₄, and concentrated. Finally, the 543 544 crude product was purified by silica gel column chromatography, employing a mixture of petroleum ether and ethyl acetate (30:1) as the eluent, yielding the desired intermediates B1 and B2. 545

546 Target compounds C1–C50 were synthesized through the same method and procedure, with 547 C1 as a representative example, a solution of N-(4-aminophenyl)-4-methoxybenzenesulfonamide (1 548 mmol) and intermediate B1 (1.2 mmol) was prepared by dissolving them in 10 mL of ethanol. A 549 catalytic quantity of glacial acetic acid was then added and then heated. The reaction progress was 550 monitored during the experiment using thin-layer chromatography. Once a significant portion of the 551 starting materials had undergone a reaction, the solvent in the mixture was concentrated. After the 552 evaporation of ethanol and subsequent aqueous workup, the obtained residue underwent purification 553 through column chromatography. This process utilized a solvent system consisting of ethyl acetate 554 and petroleum ether in a ratio of 5:1 as the eluent. This purification process ultimately yielded the target compound C1. (E) -N- (4-(((5-hydroxy-2,2-dimethyl-2H-chromen-6-yl) methylene) amino) 555 556 phenyl) -4-methoxybenzenesulfonamide (C1): Yield: 60.13%; yellow solid; m. p. 139-141 °C; ¹H 557 NMR (500 MHz, DMSO- d_6) δ 14.25 (s, 1H, -OH), 10.25 (s, 1H, -SO₂NH-), 8.76 (d, J = 4.1 Hz, 1H, -N=CH-), 7.73 – 7.67 (m, 2H, Ar-H), 7.34 – 7.26 (m, 3H, Ar-H), 7.14 (dd, J = 9.2, 3.6 Hz, 2H, Ar-558 H), 7.09 – 7.03 (m, 2H, Ar-H), 6.63 (d, J = 10.5 Hz, 1H, Ar-H), 6.38 (d, J = 7.9 Hz, 1H, Ar-H), 5.72 559 560 (d, J = 10.8 Hz, 1H, Ar-H), 3.79 (d, J = 4.2 Hz, 3H, -OCH₃), 1.39 (s, 6H, -CH₃). ¹³C NMR (126) MHz, DMSO-*d*₆) δ 163.0, 162.4, 158.6, 157.3, 143.5, 136.9, 134.1, 131.5, 129.4, 129.2, 122.3, 121.4, 561 562 116.0, 114.9, 113.3, 109.05, 108.4, 77.6, 56.1, 28.3. HRMS (ESI) m/z for C₂₅H₂₄N₂O₅S [M+H]⁺ 563 calcd 465.14787, found 465.14771.

564

565 566

567 Antiviral bioassay

information (see Data S2).

According to previous reports ⁴⁵, PVY was infested in *Nicotiana tabacum* K326, and PVY was harvested and purified after showing severe infection symptoms on the leaves. The literature-guided protocol was followed to assess the curative, prophylactic, and non-reactive potential of the target compounds under investigation ¹⁵.

The spectral profiles of all target compounds are provided in the accompanying Supporting

572 Curative Activity of Compounds Against PVY: To assess the curative effectiveness against PVY, 573 a uniform layer of 200-300 mesh emery was applied to the leaves of C. amaranticolor, followed 574 by rubbing with a virus solution. For 1.5 h, then the emery was washed off with water. After the leaves naturally dried, a 500 μ g/mL solution of the test compound was carefully applied to one side 575 576 of the leaf, while the opposite side received a 1% Tween 80 solution as a control. The plants were 577 subsequently incubated in a greenhouse maintained at 25 °C. After 5–7 d, the presence of necrotic 578 spots on the leaves was recorded, with each compound tested on three replicate plants. The average 579 inhibition rate was determined by analyzing the data from these replications.

580 Protective Activity of Compounds Against PVY: A 500 µg/mL solution of the targeted

581 compound was delicately applied to one side of *C. amaranticolor* leaves, while the other side was 582 treated with a 1% Tween 80 aqueous solution. The plants were positioned in a greenhouse set at 583 25 °C. Following a 24 h period, the entire leaf surface was inoculated with PVY by uniformly 584 distributing 200–300 mesh emery. After 1.5 h, the emery was washed off with water, and the plants 585 were returned to the greenhouse. After 5–7 d, the number of necrotic spots was tallied, with the 586 assessment based on the greenhouse population. Each compound was tested in triplicate, and the 587 average inhibition rate was subsequently calculated.

588 Inactivating Activity of Compounds Against PVY: Target compound solution (1,000 µg/mL) and 2 × virus solution were combined in equal quantities and incubated for 30 min. Evenly distribute 589 590 emery onto the leaf blades, and apply the mixture of virus and compound solution onto the right 591 side of C. amaranticolor leaf blades. Following this, an identical quantity of a 1× virus solution was 592 applied to the untreated side of the leaf. Following a duration of 1.5 h, the emery residue present on 593 the leaf surface was removed using a rinsing process utilizing water. The plants were incubated 594 within a controlled greenhouse environment, maintaining a consistent temperature of 25 °C. After 595 5-7 d, they proceeded to statistics the number of necrotic spots on the leaves. For each chemical, 596 triplicate experiments were undertaken, and the average percentage of inhibition was subsequently 597 determined.

599 **3D-QSAR studies**

600 The software SYBYL-2.1 was used to produce comparative molecular field analysis (CoMFA) 601 and comparative molecular similarity index analysis (CoMSIA) models. A total of 39 compounds were randomly chosen for training purposes, with the remaining 10 exclusively assigned for testing. 602 603 3D-OSAR models for the molecular structures of the target compounds and their effects on 604 inactivating PVY were developed using partial least squares (PLS) regression. Initially, the structures of the target compounds were subjected to energy minimization within the software. 605 606 Subsequently, compound C6 served as the reference molecule, and the 3D conformations of all 607 compounds in the training set were aligned by selecting the common scaffold. Cross-validation of 608 the training set compounds was performed using the leave-one-out strategy (LOO) diligently. This 609 process involved computing the cross-validation coefficient (q^2) and determining the optimal 610 number of principal components (ONC). The model's performance was assessed in terms of its goodness of fit, non-cross-validation coefficient (r^2), Fisher statistic (F), standard error of estimate 611 612 (SEE), and relative force field contribution. Subsequently, the model was further refined through 613 the conformational search technique. Finally, the optimized model was employed to predict the 614 biological activities of the target compounds in the test set.

615

598

616 Molecular docking and molecular dynamics simulation

617 The molecular docking was performed using the Schrödinger Maestro 13.5 program (March 618 2023 edition). The crystallographic structure corresponding to PVY CP (PDB ID: 6HXX) was initially obtained from the RCSB PDB database. The protein crystals that were obtained underwent 619 620 a series of protein preprocessing procedures, which included restoring natural ligand states, refining 621 hydrogen bond assignments, minimizing the energy of protein, and eliminating water molecules. 622 The LigPrep module was utilized to process the two-dimensional structure files of compounds C6 623 and C50, generating all conceivable three-dimensional chiral conformations. The ligand molecules 624 C6 and C50 were submitted to molecular docking with the active site of PVY CP utilizing the

maximum precision XP docking method. The lower score signifies a diminished binding free energy
 between the chemical compound and the protein, implying a heightened binding stability.

627 To enhance the binding efficacy of compound-protein complexes, we employed conventional molecular dynamics simulations through the Desmond software. The OPLS4 force field was utilized 628 629 to accurately model the interactions between the protein and small molecules, while the SPCE model 630 was adopted for the water solvent. The complex was immersed in a cubic water box and fully solvated. To ensure system neutrality, 0.150 M chloride and sodium ions were added. An initial 631 632 energy minimization was conducted using the steepest descent method, involving 50,000 steps. 633 Following this, the heavy atoms were restrained during NVT and NPT equilibration phases, 634 spanning another 50,000 steps. Throughout, the system temperature was held at 300 K and the 635 pressure at 1 bar. Upon completion of these equilibration stages, an unrestrained simulation was executed for a duration of 100 nanoseconds. The interactions were scrutinized, and dynamic 636 637 trajectory data was compiled using Maestro 13.5, facilitating further analysis and optimization of 638 the binding mode.

639

640 Plasmid construction and site-directed mutagenesis

641 The pCamPVY-GFP plasmid (GenBank accession: MN381731) was constructed in the 642 previous study ⁴⁶. Subsequently, the coding region of PVY CP was amplified via PCR from the pCamPVY-GFP and fused into a modified pCambia0390 vector, which harbors a 35S promoter and 643 644 the GFP gene, yielding the pCamGFP-PVY CP construct. For expression in Escherichia coli, the 645 PVY CP sequence was transferred into the pET-32a (+) vector, creating the pET-PVY CP plasmid. Furthermore, to introduce a Ser¹²⁵ substitution in both pCamPVY-GFP and pET-PVY CP, site-646 directed mutagenesis was employed, adhering to a previously established protocol ⁴⁷. This led to the 647 generation of pCamPVY^{S125A}-GFP and pET-PVY CP^{S125A} constructs, respectively. All the 648 649 constructed plasmids were sequenced. Table S2 lists the primers used in this study.

650

651 Agrobacterium infection of PVY-GFP, agent treatment, and photographing

The *Nicotiana benthamiana* plants were subjected to infestation by Agrobacterium tumefaciens, which carried either pCamPVY-GFP or an empty vector (mock). Following this a solution with target compound **C50** (500 μ g/mL), NNM (500 μ g/mL), and a volume equivalent to DMSO (serving as the control) was applied via spraying onto all of the leaves. At 7 dpai, the leaves of *N. benthamiana* were looked at for the GFP using a LUYOR UV light (Shanghai, China), and captured in photographs. The foliage was gathered, rapidly frozen with liquid nitrogen, and subsequently preserved in a refrigerator set at a temperature of -80 °C.

659

660 RNA extraction, cDNA synthesis, and RT-qPCR

Leaf blades of Nicotiana benthamiana were pulverized into fine powder with the aid of liquid 661 nitrogen, followed by the extraction of total RNA from the leaf tissue utilizing Trizol reagent 662 (Invitrogen, California, USA). Subsequently, cDNA synthesis was conducted through reverse 663 transcription with primers and a reverse transcription kit (Vazyme, Nanjing, China). Any potential 664 665 DNA contamination was effectively removed using a gDNA wipe enzyme (Vazyme, Nanjing, China) 666 as per the manufacturer's guidelines. RT-qPCR analysis was carried out employing TB Green 667 (TaKaRa, Beijing, China). Detailed information regarding the primers utilized in this investigation 668 can be found in Table S2.

669

670 Expression and purification of PVY CP and PVY CP mutants

The PVY CP^{wt} gene, the wild-type variant of the Potato Virus Y coat protein, was cloned into 671 the pET-32a (+) prokaryotic expression vector. This recombinant vector was then thermally 672 673 transformed into BL21(DE3) bacterial host cells for subsequent protein expression. The strains 674 above were cultivated on a liquid Luria-Bertani medium (180 rpm, 37 °C). To initiate the expression of PVY CP, isopropyl β -D-1-thiogalactopyranoside (IPTG) was added to the culture medium at a 675 676 concentration of 1.0 mmol when the optical density at 600 nm (OD₆₀₀) reached a range between 0.6 and 0.8. This step triggered the production of the PVY CP protein. The induction process was 677 maintained for 10 h (180 rpm, 16 °C). The 6 × -His-S-tagged PVY CP^{wt} were purified using nickel-678 nitriloacetic acid by high-performance column affinity chromatography. Subsequently, enterokinase 679 680 was employed to digest the proteins, and the desired target proteins were isolated using His-tag 681 purification resin. Ultimately, the proteins were identified through the employment of a 12% sodium dodecyl sulfate-polyacrylamide gel electrophoresis method. The pET-32a (+) plasmid containing 682 the PVY CP^{wt} gene was used as a template. The gene encoding serine at position 125 on PVY CP^{wt} 683 was a point mutation to the alanine gene. The mutant protein was purified using the method for the 684 purification of PVY CP^{wt}, which was PVY CP^{S125A}. 685

686

687 MST assay for the affinity of the compounds with PVY CP

The proteins' binding affinity with target chemicals was evaluated using a Monolith NT.115 instrument (Nano Temper, Munich, Germany), utilizing established methods described in literature^{35, 48}. A concentration of 2.0 mM was selected for the chemicals, and the proteins were tagged with the Monolith TM RED-NHS second-generation labeling kit. The MST assay analyzed the fluorescence intensity of the labeled proteins, focusing on a range from 400 to 1200 to investigate their interactions. The scanning method was conducted using an "LED power" setting of 40% and a temperature that was maintained at 25.0 °C.

695

696 Western blot assay

The leaves of the assayed *N. benthamiana* were used to extract total protein according to the previously described ¹⁵. An anti-GFP (Proteintech, Wuhan, China) or anti-PVY CP (Youke, Shanghai, China) antibody and horseradish peroxidase-conjugated goat anti-rabbit IgG (Proteintech, Wuhan, China) were individually used as the primary and secondary antibody. The visualization of the target protein's signal was facilitated through the use of the Chemi Doc MP Imaging System manufactured by Bio-Rad.

703

704 Virus particle purification

Leaves of *N. benthamiana* that had been infected with wild-type or mutated PVY for 5 days were harvested for the extraction of viral particles. The methodology for purifying virion particles was performed according to our established protocol. The supernatant containing PVY was absorbed onto a copper grid coated with a carbon support film (Zhongjingkeyi, Beijing, China) and subsequently stained with a 1% phosphotungstic acid buffer. Subsequently, the structure of PVY particles was visualized using transmission electron microscopy (TEM, Talos F200C, FEI, USA).

711

712 Confocal microscopy

To analyze cell-to-cell movement of PVY-GFP and its mutant in *N. benthamiana* plants, the leaf patches infiltrated with Agrobacterium carrying pCamPVY-GFP and pCamPVY^{S125A}-GFP were observed using a confocal microscope (Carl Zeiss, Germany). The excitation wavelength for observing GFP fluorescence was adjusted to 488 nm, while the emission wavelengths were specified to fall within the 520 to 540 nm range. The captured images were subsequently analyzed and processed with ZEN 2.1 software.

719

720 QUANTIFICATION AND STATISTICAL ANALYSIS

All experiments were conducted with three independent biological replicates per treatment. The sample size (n) for each measurement was determined in triplicate. Statistical analysis was performed using Microsoft Excel 2021. In **Table 1** all data are presented as means \pm Standard deviation (SD). SPSS software (version 26.0) was used for one-way ANOVA. Error bars on **Figure 3C**, **3D**, **6C** and **6F** indicate SD. Different letters indicate statistically significant differences (p < 0.05).

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8 SUPPLEMENTAL INFORMATION

729 730 Document S1. Figures S1–S3, Table S1 and S2, Data S1 and S2

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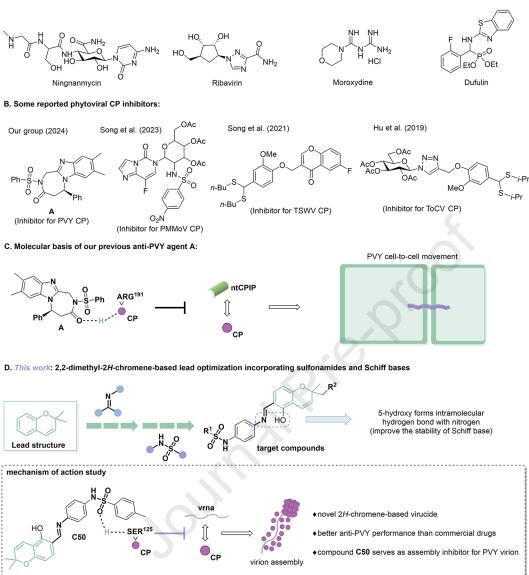
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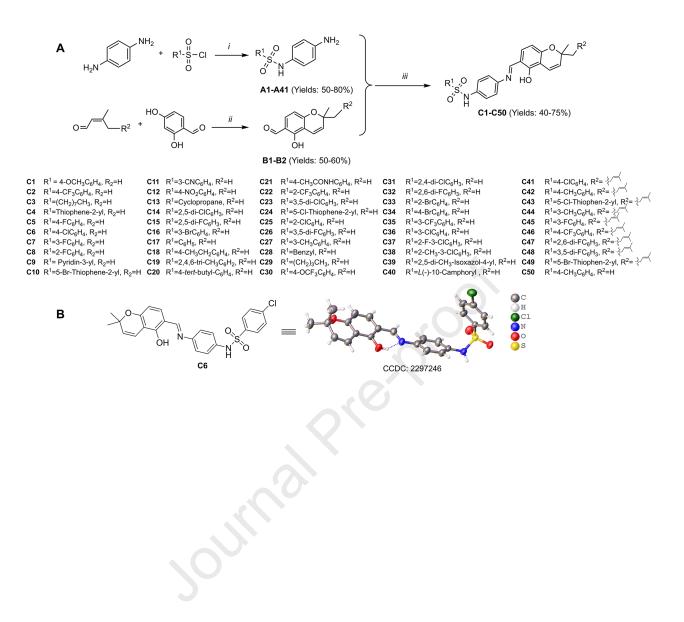
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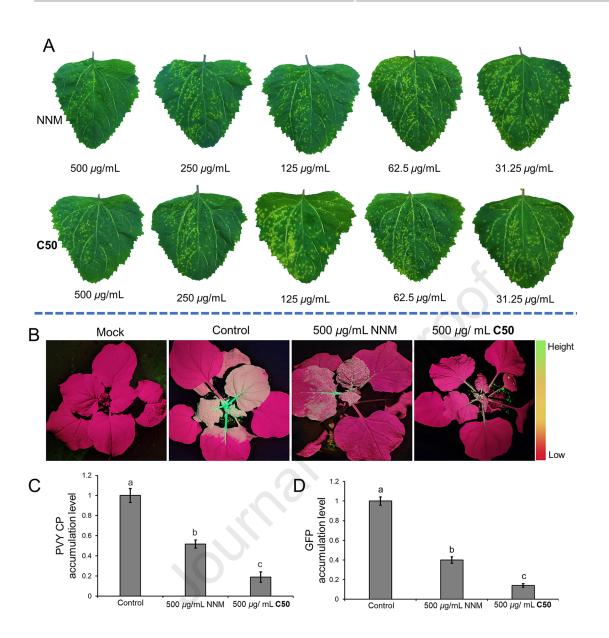
889 12, 554. 10.3390/pharmaceutics12060554.

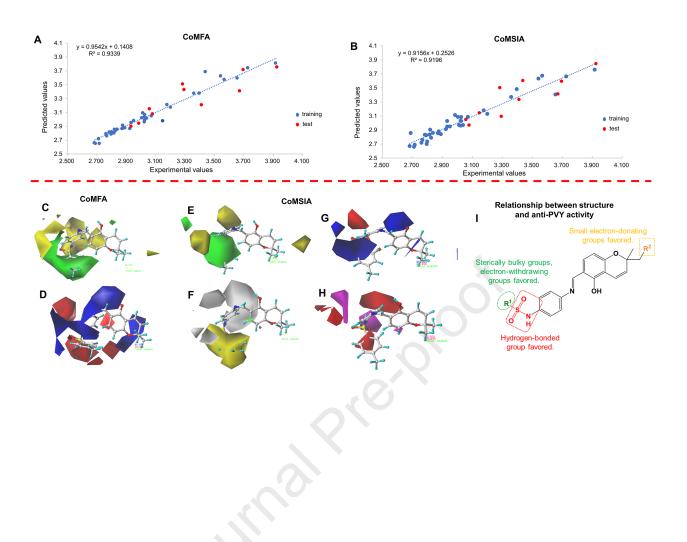
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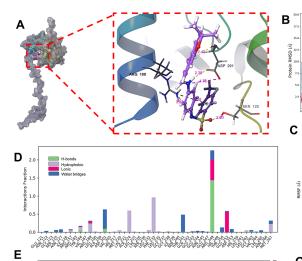
A. Active ingredients in common commercial antiviral agents for plants:

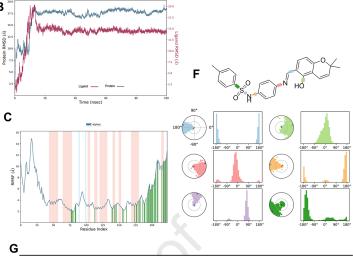




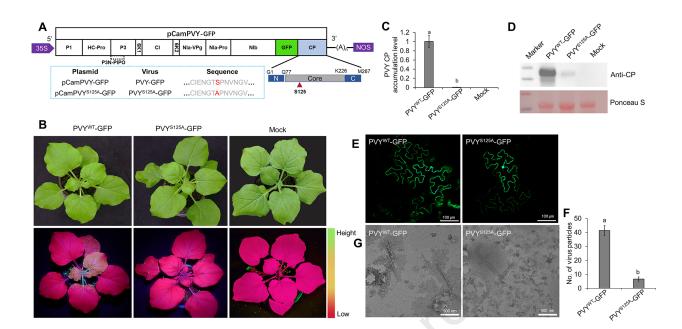


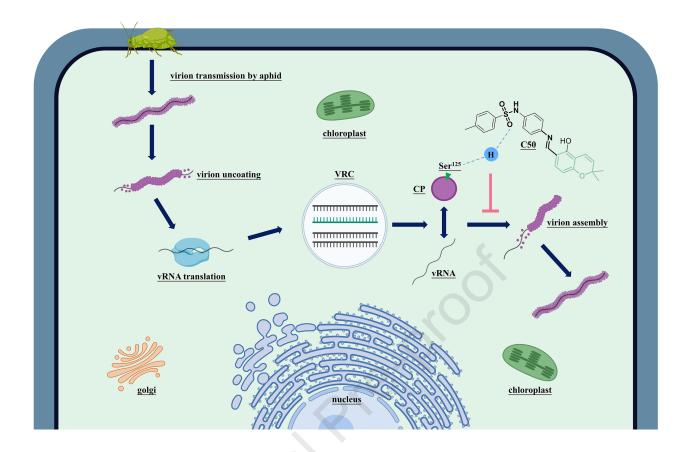






Comp.	R ¹ , O O'NHO	PVY CP ^{WT} $K_{\rm d}$ (μ M)	$\begin{array}{c} PVY \ CP^{S125A} \\ \mathcal{K}_{d} \ (\muM) \end{array}$	Inactivating activity EC ₅₀ (µg/mL)	
C6	$R^1 = 4 - CIC_6 H_4, R^2 = H$	$4.4\ \pm 1.3$	312.5 ± 40.3	56.7 ± 4.2	
C50	$R^1 = 4-CH_3C_6H_4$, $R^2 = H$	$1.6~\pm~0.3$	125.4 ± 51.8	53.3 ± 4.3	





Highlights

- 1. Discovery of 2H-chromene derivatives with potent inhibitory activity against PVY.
- 2. Constructed 3D-QASR models of 2H-chromene derivatives to guide compound optimization.
- 3. Discovery of inhibitors for PVY particle assembly based on the PVY CP conserved region.
- 4. Point mutation strategy to validate the effect of potential sites on PVY function.

Journal Pre-proof



KEY RESOURCES TABLE

REAGENT or RESOURCE	SOURCE	IDENTIFIER
Antibodies		
anti-GFP	Proteintech	Cat#66002-1-Ig
anti- PVY CP	Youke	Custom synthesis, http://www.youke- ab.cn/
Horseradish peroxidase-conjugated goat anti-rabbit IgG	Proteintech	Cat#SA00001-2
Bacterial and virus strains		
Escherichia coli DH5a Competent Cells	Sangon Biotech	Cat#B528413-0010
Escherichia coli BL21 (DE3) Competent Cells	Sangon Biotech	Cat#B528414-0010
Agrobacterium GV3101 Competent Cells	Sangon Biotech	Cat#B528430-0010
Potato virus Y	Our laboratory collection	N/A
Biological samples		
Nicotiana benthamiana	Our laboratory collection	N/A
Nicotiana tabacum K326	Our laboratory collection	N/A
Chenopodium amaranticolor	Our laboratory collection	N/A
Chemicals, peptides, and recombinant proteins		1
Isopropyl β -D-1-thiogalactopyranoside	Sangon Biotech	Cat#A600168-0005
2,4-dihydroxybenzaldehyde	Energy Chemical	Cas: 95-01-2
3-methylbut-2-enal	Energy Chemical	Cas: 107-86-8
citral	Energy Chemical	Cas: 5392-40-5
<i>p</i> -phenylenediamine	Energy Chemical	Cas: 106-50-3
Substituted sulfonyl chloride	Energy Chemical	N/A
DMSO-d ₆	Energy Chemical	Cas: 2206-27-1
Ningnanmycin	Bide Pharmatech	Cas: 156410-09-2
Ponceau S	Sangon Biotech	Cat#A610437-0025
PVY CP	Wei <i>et al</i> .(2024) ¹⁵	DOI:10.1002/advs.202 309343
PVY CP ^{S125A}	This paper	N/A
Critical commercial assays		
Trizol reagent	Invitrogen	Cat#15596026
HiScript III 1st Strand cDNA Synthesis Kit (+gDNA wiper)	Vazyme	Cat#R312-01
TB Green	TaKaRa	Cat#R820A
Monolith RED-NHS second generation protein labeling kit	NanoTemper	Cat#MO-L011
Deposited data		
Potato Virus Y coat protein structure	Kežar <i>et al.</i> (2019) ¹³	PDB (entry 6HXX)
(<i>E</i>)-4-chloro- <i>N</i> -(4-(((5-hydroxy-2,2-dimethyl-2 <i>H</i> -chromen-6-yl)methylene)amino)phenyl) benzenesulfonamide crystallographic data	This paper	CCDC No. 2297246
Oligonucleotides		
Primers used for DNA amplification, RT-qPCR and mutagenesis	This paper	See Supplementary information (Table S2)
Recombinant DNA		
pCAMBIA0390 expression vector for PVY-GFP	Cheng <i>et al.</i> (2020) ⁴⁶	DOI:10.1016/j.virusres .2019.197827
pCAMBIA0390 expression vector for PVY ^{S125A} -GFP	This paper	N/A



pET32a (+) expression vector for PVY CP	Sangon Biotech	Custom synthesis, https://www.sangon.co m
pET32a (+) expression vector for PVY CP ^{S125A}	Sangon Biotech	Custom synthesis, https://www.sangon.co m
Software and algorithms		
NanoTemper Monolith Instrument (NT.115) Control and Analysis Software Package	NanoTemper	https://nanotempertech .com/microscale- thermophoresis
SYBYL-2.1	Tripos	https://www.tripos.co m
Schrödinger Maestro 13.5	Schrödinger	https://www.schroding er.com
MestReNova	Mestrelab Research	https://www.mestrelab cn.com
SPSS 26.0	IBM	https://www.ibm.com
Microsoft Excel 2021	Microsoft	https://www.microsoft. com
ZEN 2.1	ZEISS	https://www.zeiss.com
ChemDraw 23	CambridgeSoft	https://www.chemdraw .com.cn
Chemi Doc MP Imaging System	Bio-Rad	https://www.bio- rad.com
Desmond	Schrödinger	https://www.schroding er.com
Other		
Copper mesh carbon support film	Zhongjingkeyi	Cat#BZ11022a